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*phET Simulation
on Molecule*

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~~Simulation~~

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Molecule

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Molecule Shapes

Lab - Build a

Molecule ~~Module~~

~~3.2: Bonding~~

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~~Polar Molecules
Tutorial: How to
determine
polarity in a
molecule~~

Molecule

Polarity Polar

\u0026 Non-Polar
Molecules: Crash
Course Chemistry

#23 AP

~~Chemistry:~~

~~3.11-3.13~~

~~Spectroscopy,~~

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~~Photoelectric
Effect, and Beer
Lambert Law Elec
tronegativity~~

and Bond

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Simulation -

Revision for A-

Level Chemistry

Building a

Molecule Density

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Help with

Worksheet

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*Polarity Rules
Molecule
Polarity -
Chemistry
Tutorial*

VSEPR Theory:
Introduction
~~Intermolecular
Forces and
Boiling Points~~
*How to Determine
if a Molecule is
Polar or Not*

Bonding-polar vs
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nonpolar
molecules

-keeping it
simple *Polar and
Nonpolar*

*Covalent Bonds -
Clear \u0026*

Simple Polarity

4.3 Predict

molecular
polarity from
molecular shape
and bond

polarities [SL

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IB Chemistry]4.5

~~Polarity~~

AP Chemistry:

3.4-3.6 Ideal

Gas Law and

Kinetic

Molecular Theory

~~Explain why a~~

~~simple covalent~~

~~compound is a~~

~~gas at room temp~~

~~but a giant~~

~~covalent is a~~

~~solid~~ 021220

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Nomenclature and
VSEPR

Integrating
Computational
Science into the
Chemistry

Curriculum ~~The~~
~~whole of OCR~~
~~gateway~~

~~Chemistry topic~~

~~6 — Global~~

~~Challenges GCSE~~

~~Revision~~

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Lab Answer Key

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Molecule

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Answers
Lab A study of e
lectronegativity

, bond polarity,
and molecular
polarity

Introduction: In
this atomic-
level

simulation, you
will investigate
how atoms' elect
ronegativity

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value affects the

Simulation

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Molecule

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Lab Answer Key

The molecule is polar because it does not have the same charge.

H=Positive charge.

F=Negative charge. H₂O

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(water) The molecule is polar, because it doesn't have the same charge. H1 and H2=Have a positive and an electrostatic potential charge.

O=Negative charge. CH₂F₂ Is polar because, it have some

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charges that
have atoms that
are positive and
negative.

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Molecule
Polarity In this
activity you
will ...**

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Molecular
Polarity 1. Go
to <https://phet.c>

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olorado.edu/en/s
imulaAon/molecul
e-polarity.

Click on the
picture of the
simulaAon to
launch it. Part
1 (*Make sure to
take at least
one screenshot
of this part!)

2. Choose Two
Atoms. 3. Under
View, click the

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box to show
Partial Charges.
4. Without
changing
anything else,
answer the
following
questions: a.

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simulations
Lab.pdf - CHEM
115 Lab 9
Molecular ...**

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When is a molecule polar? Change the electronegativity of atoms in a molecule to see how it affects polarity. See how the molecule behaves in an electric field. Change the bond angle to see how shape affects

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Simulation

Molecule

Polarity -

**Polarity | Elect
ronegativity -**

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Molecule

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Lab Answer Key A

whole molecule

may also have a

separation of

charge,

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depending on its
molecular
structure and
the polarity of
each of its
bonds If such a
charge
separation
exists, the
molecule is said
to be a polar
molecule (or
dipole); . 17 The
molecular

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Structure of the
methane
molecule, CH₄,
is shown with a
tetrahedral ...

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Answers atom
molecule. 3.
Record your
ideas in the
table below.

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Change the bond
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close. Embed a
running copy of
this simulation.
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ps://phet.colora

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/molecule-shapes
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Pairs | Bonds -
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PhET sims are based on extensive education research and engage students through an intuitive, game-like environment where students learn through exploration and discovery.

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Shapes Lab.

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lectronegativity

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and molecular

polarity

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Simulation, you will investigate how atoms' electronegativity value affects the bonds they produce. When two atoms bond, a pair of electrons is shared between atoms.

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