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students of IIT Madras in the July–December 2009 semester.

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The kinetic and the potential energy are negligible in



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many chemical reaction engineering applications, so Equation 15 becomes  $Q - V E = Q - V (U + K + E + P) \dot{V} = Q - V U$  (16) we know that  $U$  is a function of the enthalpy, pressure, and volume, so  $Q - V U = Q - V (H - PV) = Q - V \sum_{i=1}^n X_i n_i h_i - P Q = \sum_{i=1}^n X_i n_i F_i h_i - P Q$  (17) When this is applied for both streams, the term  $PQ$  cancels with the volumetric work

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ChE471: CHEMICAL REACTION ENGINEERING (Fall 2012) Lecture in Green L0159 Instructor: Professor Milorad Dudukovic (dudu@wustl.edu). Teaching Assistant: Tim Boungh Wook Lee (bounghwooklee@go.wustl.edu) Office Hours 1-2 PM Wednesdays in Brauer 1050

## **ChE471: Chemical Reaction Engineering**

1 Chapter 1: Introduction to Separation Process Engineering Why are we—as chemical engineers—required to study “separation processes”? Separations are crucial in chemical engineering (e.g., chemical plants, petroleum refineries) Chemical

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plants commonly have from 40% to 70% of both capital and operating costs in separations

## **01 Introduction to Separation Process Engineering**

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