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~~#13: The~~

~~Equilibrium~~

~~Constant. Le~~

~~Chatelier Lab~~

~~ANSWERS:  $Fe^{3+}$~~

~~and  $FeSCN^{2+}$~~

~~Equilibrium~~

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Determination of

$K_{eq}$  for  $FeSCN^{2+}$

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### Lab Explanation

#### Video

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Le Chatelier's  
Principle Lab  
with Cobalt  
Complex Ions  
*Equilibrium*  
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*Stoichiometry*  
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~~Equilibrium~~  
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~~chemical~~

~~reactions (and~~

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~~Aaron Sams~~

Chapter 15

Chemical

Equilibrium

*Keq FeSCN<sup>2+</sup> Lab*

*Equilibrium*

*Constant*

**Determining an**

**Equilibrium**

**Constant by Spec**

**trophotometry**

**Procedure**

*Equilibrium*



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Part 1 What is  
Density? |

Gravitation |

Physics | Don't

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~~PHYSICAL~~

~~PROCESSES~~

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Experiment on

First Condition

of Equilibrium

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Physics Pre-Lab:

Experiment #2

First Condition

of Equilibrium

~~Equilibrium~~

~~animation~~ Flinn

At-Home Lab

8-Types of

Chemical

Reactions -

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*Crisis -*

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~~Hall Seattle~~

~~Physics Vs~~

~~Engineering |~~

~~Which Is Best~~

~~For You? Stefan~~

*Wager:*

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*Experimenting in*

*Equilibrium* Why

does ice float

in water? -

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**Principle Lab**

**Video**

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What is Force? -

Part 1 | Forces

and Motion |

Physics | Don't

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Analysis

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Lab Lab Write-

up: Answers to

*Page 17/48*

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### Equilibrium Lab

questions, graphs, data table. Group.

PART A. Login to the computer and open a web browser. Go to. Click "Play with Sims," then find the "Chemistry" section, and choose "Reactions and Rates."

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Pre-Lab

Questions 1.

False, The

reactions

continue, but

there is an

equal balance of

opposing

reaction rates

2. False, the

concentration of

reactants and

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Equilibrium Lab

products remain constant with time. They don't necessarily have to be equal. 3.

(a) When the humidity is low, the paper will be blue.

**Color lab -**

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**Equilibrium Lab**

**Pre-Lab**

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Questions 1 . . .

Answers  
Founded in 2002  
by Nobel

Laureate Carl  
Wieman, the PhET  
Interactive  
Simulations  
project at the  
University of  
Colorado Boulder  
creates free  
interactive math  
and science  
simulations.

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PhET Equilibrium Lab

Answers  
based on  
extensive

education <a  
{0}>research</a>  
and engage  
students through  
an intuitive,  
game-like  
environment  
where students  
learn through  
exploration and  
discovery.



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Lab Part A -  
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The amount of  
reactants

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Answers  
decreases and  
the amount of  
products

increases in  
order to reach  
equilibrium. b)

Describe the  
nature of  
dynamic

equilibrium when  
small numbers of  
particles (such  
as 50, as  
compared to

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6.022 x 1023 ) Lab

are present.

When small amounts of particles are present, equilibrium is not reached immediately; rather, it is a slow process. However, the reaction progresses at a

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### Equilibrium, steady rate.

## Answers

### Equilibrium Lab | Chemical Equilibrium | Reaction Rate

1. When red-colored  $\text{Co}(\text{NO}_3)_2 \cdot 6\text{H}_2\text{O}$  crystals are dissolved in 3mL of deionized water, the

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Equilibrium Lab

Answers  
solution turns pink. To explain this, the

equilibrium

stress is on the product's side

(addition of

water), so the

solution shifts

towards to

reactants. Since

the reactants

are favored, the

2.

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Equilibrium Lab  
**Lab 5- Chemical  
Equilibrium and  
Le Chatelier's  
Principle ...**

In this lab, the effect of applying stresses to a variety of chemical systems at equilibrium will be explored. The

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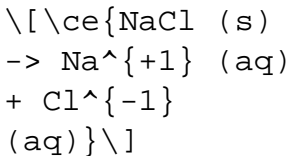
Equilibrium Lab

systems to be  
studied are

given below:

Saturated Sodium  
Chloride

Solution



Acidified

Chromate

Solution

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### Equilibrium Lab

**12: Equilibrium  
and Le**

**Chatelier's**

**Principle**

**(Experiment ...**

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Equilibrium: It

Works Both Ways

Student

Laboratory Kit

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The word equilibrium has two roots: aequ, meaning equal, and libra, meaning weight or balance. Our physical sense of equilibrium suggests a condition of equal balance of opposing forces

How does this

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Answers  
of equilibim

translate to

chemical

equilibrium? The

purpose of this

lab is to

explore the

nature and

consequences of

equilibrium in a

varicty of

chenical

neactions

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Chemical Equilibrium Lab  
Concepts .  
Answers

**Solved:**  
**Exploring**  
**Equilibrium: It**  
**Works Both Ways**  
**Student ...**

Describe the  
relative sizes  
of the forward  
and reverse  
rates at  
equilibrium.

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Answers  
Explain what effects whether the equilibrium position favors the products or the reactants.

Predict how addition of a reactant or product will affect the forward and reverse reaction rates, and once

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Exploring

this new system Lab

reaches

equilibrium how

the reactant and

product

concentrations

will compare to

the original

system at ...

**Reactions &**

**Rates - Reaction**

**| Kinematics |**

**Concentration**

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### Equilibrium Lab

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Report On

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Equilibrium It

Works Both Ways

Solution  $\text{NaCl}(s)$

$\text{Na} + \text{Cl}(aq) \rightleftharpoons \text{NaCl}(aq)$

?  $\text{Cl}(aq)$  12:

Equilibrium and

Le Chatelier's

Principle

(Experiment...

Equilibrium is a



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### Equilibrium in Lab

Answers

condition in  
which all the  
forces in a  
system

counteract each other. When adding vectors, equilibrium can be showed by a representative force, called an equilibrant. The

**Lab Report On**

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**Exploring Equilibrium Lab**

**Equilibrium It**

**Works Both Ways**

Here you will  
test the effects  
of changing the  
volume and  
temperature on  
the complex ion  
equilibrium

between Co

$(\text{H}_2\text{O})_6^{2+} (\text{aq})$

and  $\text{CoCl}_4^{2-} (\text{aq})$

as in Reaction

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### 3.14. Reagents Lab

needed for this part are:  $\text{CoCl}_2 \cdot 6\text{H}_2\text{O}$  (s), 12 M  $\text{HCl}$  (aq) (do not remove from fume hood), and deionized water. Reheat your hot-water bath from part B to a near boil.

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**Principle**

**(Experiment) -**

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